

Heat Transfer

- Heat is the energy transferred from one system to another or from one part of the system to its other part, arising due to temperature difference.
- Heat can be transferred from one place to other by through three different modes conduction, convection and radiation.
- **Thermal Conduction** of heat takes place in a body when different parts of body are at different temperature.

- **Heat Conduction**

$$Q = -KA \left(\frac{dT}{dx} \right)$$

- Convection is transfer of heat by actual motion of matter.
- Radiation process does not need any material medium for heat transfer.
- Term Radiation refers to the continuous emission of energy from surface of all bodies and this energy is called radiant energy.

- **Wein displacement law**

$$\lambda T = b$$

where $b = 0.2896 \times 10^{-2}$ mK for black body and is known as Wien's constant.

- **Stefan Boltzmann law**

$$u = e\sigma T^4$$

u is positive if net energy is being absorbed via radiation and negative if it is being lost via radiation.

- **Newton law of Cooling**

For small temperature difference between the body and surrounding rate of cooling is directly proportional to the temperature difference and surface area exposed i.e.,

$$\frac{dT}{dt} = b(T - T_s)$$

b depends on nature of surface involved and the surrounding conditions.

Negative sign is to indicate that $T_1 > T_2$, dT/dt is negative and temperature decreases with time